# Safety Data Sheet



According to the UN GHS revision 8

Creation Date: May 15, 2024 May 15, 2024 Revision Date:

### **IDENTIFICATION**

#### 1.1 **GHS Product identifier**

**Product name:** 4-Nitrobenzoic acid

Catalog Number: T20882 **CAS Number:** 62-23-7

#### 1.2 Other means of identification

Other names:

#### Recommended use of the chemical and restrictions on use 1.3

**Identified uses:** 

#### Supplier's details 1.4

Company: Targetmol Chemicals Inc.

Uses advised against: 36 Washington Street, Wellesley Hills, Massachusetts 02481 USA

Tel/Fax: (781) 999-4286

#### 1.5 **Emergency phone number**

**Emergency phone number:** 781-999-4286

Monday to Friday, 9am-5pm (Standard timezone: UTC/GMT -5hours). Service hours:

#### **HAZARD IDENTIFICATION** 2.

#### 2.1 Classification of the substance or mixture

Acute toxicity - Category 4, Oral Eye irritation, Category 2

#### 2.2 GHS label elements, including precautionary statements

Pictogram(s):

Signal word: Warning

H302 Harmful if swallowed Hazard statement(s): H319 Causes serious eye irritation

Precautionary statement(s):

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

Prevention: P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing

protection/...

P301+P317 IF SWALLOWED: Get medical help.

P330 Rinse mouth.

Response: P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses,

if present and easy to do. Continue rinsing.

Storage: none

Disposal: P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance

Page 1 of 7 www.targetmol.com with applicable laws and regulations, and product characteristics at time of disposal.

# 2.3 Other hazards which do not resultin classification

no data available

#### 3. COMPOSITION/INFORMATION ON INGREDIENTS

#### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number
4-Nitrobenzoic acid	-	62-23-7	200-526-2

#### 4. FIRST-AID MEASURES

# 4.1 Description of necessary first-aid measures

#### General advice

no data available

#### If inhaled

Fresh air, rest.

#### Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap.

#### Following eye contact

Rinse with plenty of water (remove contact lenses if easily possible).

### **Following ingestion**

Rinse mouth. Refer for medical attention.

# 4.2 Most important symptoms/effects, acute and delayed

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR if necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on the left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. Poisons A and B

# 4.3 Indication of immediate medical attention and special treatment needed, if necessary

SYMPTOMS: Symptoms of exposure to this compound include irritation of the skin, eyes, mucous membranes and upper respiratory tract. ACUTE/CHRONIC HAZARDS: This compound may be harmful by inhalation, ingestion or skin absorption. It is an irritant of the skin, eyes, mucous membranes and upper respiratory tract. When heated to decomposition it emits toxic fumes of carbon monoxide, carbon dioxide and nitrogen oxides. (NTP, 1992)

#### 5. FIRE-FIGHTING MEASURES

### 5.1 Extinguishing media

Fires involving this material can be controlled with a dry chemical, carbon dioxide or Halon extinguisher. A water spray may also be used. (NTP, 1992)

# 5.2 Specific hazards arising from the chemical

This chemical is combustible. (NTP, 1992)

### 5.3 Special protective actions for fire-fighters

Use foam, dry powder, carbon dioxide.

# 6. ACCIDENTAL RELEASE MEASURES

# 6.1 Personal precautions, protective equipment and emergency procedures

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Sweep spilled substance into covered containers.

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# 6.2 Environmental precautions

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Sweep spilled substance into covered containers.

# 6.3 Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

#### 7. HANDLING AND STORAGE

# 7.1 Precautions for safe handling

NO open flames. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

# 7.2 Conditions for safe storage, including any incompatibilities

Separated from strong oxidants, bases and strong reducing agents.

# **EXPOSURE CONTROLS/PERSONAL PROTECTION**

# 8.1 Control parameters

8.

#### Occupational Exposure limit values

MAK: (inhalable fraction): 4 mg/m3; peak limitation category: I(2); carcinogen category: 3B; pregnancy risk group: C

#### **Biological limit values**

no data available

# 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

# 8.3 Individual protection measures, such as personal protective equipment (PPE)

# Eye/face protection

Wear safety spectacles.

# Skin protection

Protective gloves.

#### Respiratory protection

Use local exhaust.

#### Thermal hazards

no data available

# 9. PHYSICAL AND CHEMICAL PROPERTIES

**Physical state** Solid. Crystalline.

**Color** Colorless, Yellow-white.

**Odour** no data available

Melting point/ freezing point Ca. 238 °C.

Boilingpoint or initial boiling point

and boiling range

96°C/10mmHg(lit.)

Flammability Combustible. Gives off irritating or toxic fumes (or gases) in a fire.

Lower and upper explosion

limit/flammability limit

no data available

Flash point 237°C

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Auto-ignition temperature 300°C

**Decomposition temperature** 350°C

**pH** Ca. 2.8.

Kinematic viscosity no data available

**Solubility** DMSO: Soluble,

N-octanol-water partition

coefficient

log Pow = Ca. 1.89. Remarks:The temperature and pH is unknown.

Vapour pressure Ca. 0 mm Hg. Temperature: Ca. 25 °C.

**Density and/ or relative density** Ca. 1.6.

**Relative vapour density** no data available

Particle characteristics no data available

#### 10. STABILITY AND REACTIVITY

# 10.1 Reactivity

Reacts with bases, reducing agents and strong oxidants strong oxidants. Decomposes on heating and on burning. This produces toxic fumes including nitrogen oxides nitrogen oxides.

# 10.2 Chemical stability

no data available

# 10.3 Possibility of hazardous reactions

CombustibleDust explosion possible if in powder or granular form, mixed with air. If dry, it can be charged electrostatically by swirling, pneumatic transport, pouring, etc.P-NITROBENZOIC ACID is incompatible with strong oxidizers. It is also incompatible with strong bases (potassium hydroxide). It may react with cyanides. (NTP, 1992)

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### 10.4 Conditions to avoid

no data available

# 10.5 Incompatible materials

Mixtures of the acid with potassium hydroxide (1:2 mol) readly deflagrated,

# 10.6 Hazardous decomposition products

When heat to decomposition it emits toxic fumes of /nitrogen oxides/

# 11. TOXICOLOGICAL INFORMATION

# Acute toxicity

Oral: no data available Inhalation: no data available Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

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no data available

#### **Reproductive toxicity**

no data available

#### STOT-single exposure

The substance is irritating to the eyes, respiratory tract and skin.

#### STOT-repeated exposure

Animal tests show that this substance possibly causes toxicity to human reproduction or development.

#### Aspiration hazard

A nuisance-causing concentration of airborne particles can be reached quickly when dispersed, especially if powdered.

# 12. ECOLOGICAL INFORMATION

# 12.1 Toxicity

Toxicity to fish: LC50 Brachydanio rerio (Zebrafish) >500 mg/L/48-hr; static bioassay (OECD guideline 203) Toxicity to daphnia and other aquatic invertebrates: LC50 - Daphnia magna - ca. 1 295.053 mg/L - 48 h. Toxicity to algae: EC5 - 537.796 mg/L - 96 h. Toxicity to microorganisms: no data available

# 12.2 Persistence and degradability

AEROBIC: Decomposition of nitrobenzoic acid took greater than 64 days by a soil microflora innoculum in mineral salts medium(1). After 180 minutes, little oxygen consumption by phenol adapted biological cultures occurred with 4-nitrobenzoic acid(2). The amounts of oxygen consumed after the 180 minute test time are 55 uL (endogenous), 64 uL (cells plus 100 mg/L 3-nitrobenzoic acid), and 346 uL (cells plus phenol - after 90 minutes) which results in a ratio of only 1.2 endogenous to 4-nitrobenzoic acid oxygen consumption(2). In a 2 week Japanese MITI test using 100 mg/L 4-nitrobenzoic acid and 30 mg/L sludge, 4-nitrobenzoic had a theoretical BOD of 62%(3). 4-Nitrobenzene reached 50.2% of its BOD in river water collected from Songhua River, China after 5 days. The river water samples contained about 8.0 mg/L dissolved oxygen, 800-3000/mL bacteria counts, pH 6.8-7.0, and temperatures of 15-20 deg C(4).

# 12.3 Bioaccumulative potential

An estimated BCF of 3.2 was calculated for 4-nitrobenzoic acid(SRC), using a log Kow of 1.89(1) and a regression-derived equation(2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organisms is low(SRC).

# 12.4 Mobility in soil

The Koc of 4-nitrobenzoic acid is estimated as 250(SRC), using a log Kow of 1.89(1) and a regression-derived equation(2). According to a classification scheme(3), this estimated Koc value suggests that 4-nitrobenzoic acid is expected to have moderate mobility in soil. The pKa of 4-nitrobenzoic acid is 3.44(4), indicating that this compound will exist almost entirely in the anion form in the environment and anions generally do not adsorb more strongly to soils containing organic carbon and clay than their neutral counterparts(5).

# 12.5 Other adverse effects

no data available

# 13. DISPOSAL CONSIDERATIONS

# 13.1 Disposal methods

# **Product**

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

# Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

#### 14. TRANSPORT INFORMATION

# 14.1 UN Number

no data available

### 14.2 UN Proper Shipping Name

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no data available

# 14.3 Transport hazard class(es)

no data available

# 14.4 Packing group, if applicable

no data available

#### 14.5 Environmental hazards

no data available

# 14.6 Special precautions for user

no data available

# 14.7 Transport in bulk according to IMO instruments

no data available

15.

# REGULATORY INFORMATION

# 15.1 Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)	Listed.
EC Inventory	Listed.
United States Toxic Substances Control Act (TSCA) Inventory	Listed.
China Catalog of Hazardous chemicals 2015	Not Listed.
New Zealand Inventory of Chemicals (NZIoC)	Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)	Listed.
Vietnam National Chemical Inventory	Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)	Listed.
Korea Existing Chemicals List (KECL)	Listed.

# 16. OTHER INFORMATION

# Information on revision

Creation Date May 15, 2024

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### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.

org/echemportal/index?pageID=0&request\_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

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ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp ECHA - European Chemicals Agency, website: https://echa.europa.eu/

### **Other Information**

no data available

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