

According to the UN GHS revision 8

Creation Date: April 20, 2026

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1. IDENTIFICATION

1.1 GHS Product identifier

Product name: (-)-CAMPHOR

Catalog Number: TMS2262

CAS Number: 464-48-2

1.2 Other means of identification

Other names: -

1.3 Recommended use of the chemical and restrictions on use

Identified uses:

1.4 Supplier's details

Company: Targetmol Chemicals Inc.

Address: 34 Washington Street, Wellesley Hills, Massachusetts 02481 USA

Tel/Fax: (781) 999-4286

1.5 Emergency phone number

Emergency phone number: 781-999-4286

Service hours: Monday to Friday, 9am-5pm (Standard timezone: UTC/GMT -5 hours).

2. HAZARD IDENTIFICATION

2.1 Classification of the substance or mixture

Germ cell mutagenicity, Category 2

2.2 GHS label elements, including precautionary statements

Pictogram(s):



Signal word: Warning

Hazard statement(s): H341 Suspected of causing genetic defects

Precautionary statement(s):

Prevention: P203 Obtain, read and follow all safety instructions before use.
P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

Response: P318 IF exposed or concerned, get medical advice.

Storage: P405 Store locked up.

Disposal: P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

2.3 Other hazards which do not result in classification

no data available

3. COMPOSITION/INFORMATION ON INGREDIENTS

3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number
(-)-CAMPHOR	-	464-48-2	207-354-7

4. FIRST-AID MEASURES

4.1 Description of necessary first-aid measures

General advice

no data available

If inhaled

Fresh air, rest. Artificial respiration may be needed. Refer for medical attention.

Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower.

Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

Following ingestion

Rinse mouth. Give a slurry of activated charcoal in water to drink. Artificial respiration may be needed. Refer for medical attention .

4.2 Most important symptoms/effects, acute and delayed

Treatment of camphor intoxication is primarily supportive with a focus on airway management and seizure control. No antidotes are available. Activated charcoal should be administered for gastrointestinal decontamination, although its efficacy is doubtful. Due to prominent CNS effects, the induction of emesis is contraindicated. If liquid camphor is ingested, a nasogastric tube can be used to aspirate gastric contents before instillation of activated charcoal. Alcohols and oil solutions should be avoided because they have been reported to enhance absorption of camphor. Although not readily available, lipid hemodialysis and resin hemoperfusion have been reported to lower blood camphor concentrations in severely poisoned patients. Benzodiazepines such as lorazepam or diazepam are indicated for symptoms of CNS hyperactivity, such as agitation, tremors, and seizures. Phenobarbital can be used for recurrent or prolonged seizures.

4.3 Indication of immediate medical attention and special treatment needed, if necessary

SYMPTOMS: Ingestion of this compound may cause nausea, vomiting, vertigo, mental confusion, delirium, convulsions, coma, respiratory failure or death. It may also cause a burning sensation, coughing, wheezing, laryngitis, shortness of breath, headache, severe irritation and possible destruction to the tissues of the mucous membranes, upper respiratory tract, eyes and skin. Other symptoms may include congestion and edematous changes in the gastrointestinal tract, kidneys and brain. Ingestion may result in burning in the mouth and throat, epigastric pain, thirst, feeling of tension, dizziness, irrational behavior, unconsciousness, rigidity, rapid pulse, slow respiration, twitching of the facial muscles and muscular spasms. Other symptoms may include flickering, darkening or veiling of vision, noises in the ears and weakness. Exposure to this compound may also result in a feeling of warmth, depression of the central nervous system, difficult breathing, a characteristic breath odor and anuria. Colic may also be a symptom of exposure. Other symptoms may include eye irritation, sore throat, excitement, fever, bluish lips, pale face, loss of sense of smell and agitation. **ACUTE/CHRONIC HAZARDS:** When heated to decomposition this compound emits toxic fumes of carbon monoxide and carbon dioxide. It is harmful if swallowed, inhaled or absorbed through the skin. It can be absorbed through mucous membranes. (NTP, 1992) Within 5 to 90 minutes after swallowing, the following may be noted: nausea and vomiting; feeling of warmth; headache; confusion, vertigo, excitement, restlessness, delirium, and hallucinations; increased muscular excitability, tremors, and jerky movements; epileptiform convulsions, followed by depression (convulsions sometimes occur early in the syndrome and may be severe, but they do not have the grave prognosis of strychnine convulsions); coma; central nervous depression may at times be the primary clinical response; death results from respiratory failure or from status epilepticus; slow convalescence (days or weeks), often with persistent gastric distress. (USCG, 1999) Excerpt from ERG Guide 133 [Flammable Solids]: Fire may produce irritating and/or toxic gases. Contact may cause burns to skin and eyes. Contact with molten substance may cause severe burns to skin and eyes. Runoff from fire control may cause pollution. (ERG, 2016)

5. FIRE-FIGHTING MEASURES

5.1 Extinguishing media

To fight fire, use foam, carbon dioxide, dry chemical

5.2 Specific hazards arising from the chemical

This chemical is flammable. (NTP, 1992) Behavior in Fire: The solid often evaporates without first melting. (USCG, 1999) Excerpt from ERG Guide 133 [Flammable Solids]: Flammable/combustible material. May be ignited by friction, heat, sparks or flames. Some may burn rapidly with flare-burning effect. Powders, dusts, shavings, borings, turnings or cuttings may explode or burn with explosive violence. Substance may be transported in a molten form at a temperature that may be above its flash point. May re-ignite after fire is extinguished. (ERG, 2016)

5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

6. ACCIDENTAL RELEASE MEASURES

6.1 Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

6.2 Environmental precautions

Personal protection: filter respirator for organic gases and particulates adapted to the airborne concentration of the substance. Ventilation. Remove all ignition sources. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting.

6.3 Methods and materials for containment and cleaning up

Methods and materials for containment and cleaning up: sweep up and shovel. Contain spillage, and then collect with an electrically protected vacuum cleaner or by wet-brushing and place in container for disposal according to local regulation... Keep in suitable, closed containers for disposal...

7. HANDLING AND STORAGE

7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

7.2 Conditions for safe storage, including any incompatibilities

Separated from strong oxidants, strong reducing agents, chlorinated solvents and food and feedstuffs. Well closed. Ventilation along the floor. Conditions for safe storage, including any incompatibilities: keep container tightly closed in a dry and well-ventilated place.

Pure form: -20°C for 3 years | In solvent: -80°C for 1 year

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

8.1 Control parameters

Occupational Exposure limit values

Recommended Exposure Limit: 10 Hour Time-Weighted Average: 2 mg/cu m.

Biological limit values

no data available

8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

Skin protection

Wear fire/flare resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

Thermal hazards

no data available

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical state	Viscous
Color	White
Odour	Fragrant and penetrating odor
Melting point/freezing point	174°C(lit.)
Boiling point or initial boiling point and boiling range	129°C/15mmHg(lit.)
Flammability	Combustible Solid
Lower and upper explosion limit/flammability limit	Lower flammable limit: 0.6% by volume; Upper flammable limit: 3.5% by volume
Flash point	65°C
Auto-ignition temperature	871° F (NTP, 1992)
Decomposition temperature	no data available
pH	no data available
Kinematic viscosity	no data available
Solubility	DMSO: 50 mg/mL (328.45 mM),Sonication is recommended. (< 1 mg/ml refers to the product slightly soluble or insoluble)
N-octanol-water partition coefficient	no data available
Vapour pressure	0.225mmHg at 25°C
Density and/or relative density	0.990 g/cm ³
Relative vapour density	5.24 (NTP, 1992) (Relative to Air)
Particle characteristics	no data available

10. STABILITY AND REACTIVITY

10.1 Reactivity

Decomposes on burning. This produces toxic gases and irritating fumes. Reacts violently with strong oxidants, strong reducing agents and chlorinated solvents. This generates fire and explosion hazard.

10.2 Chemical stability

no data available

10.3 Possibility of hazardous reactions

Evolves flammable and explosive vapors when heated.Dust explosion possible if in powder or granular form, mixed with air.L-CAMPHOR may be sensitive to heat and direct sunlight. Incompatible with strong oxidizing agents, strong reducing agents and chlorinated solvents. Also incompatible with potassium permanganate. Salts of any kind should not be added to it in water. Reacts violently with chromic anhydride (NTP, 1992).

10.4 Conditions to avoid

no data available

10.5 Incompatible materials

Reacts violently with ...strong reducing agents and chlorinated solvents, causing fire and explosion hazard.

10.6 Hazardous decomposition products

The substance decomposes on burning producing toxic gases and irritating fumes.

11. TOXICOLOGICAL INFORMATION

Acute toxicity

Oral: LD50 Mouse oral 1310 mg/kg

Inhalation: no data available

Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

A4; Not classifiable as a human carcinogen. Camphor, synthetic

Reproductive toxicity

no data available

STOT-single exposure

no data available

STOT-repeated exposure

no data available

Aspiration hazard

no data available

12. ECOLOGICAL INFORMATION

12.1 Toxicity

Toxicity to fish: LC50; Species: Pimephales promelas (Fathead minnow); Conditions: static bioassay; Concentration: 145 mg/L for 1 hr; 112 mg/L/24 hr; 111 mg/L/48 hr; 110 mg/L/72 hr; 110 mg/L/96 hr

Toxicity to daphnia and other aquatic invertebrates: no data available

Toxicity to algae: no data available

Toxicity to microorganisms: no data available

12.2 Persistence and degradability

AEROBIC: Camphor, at an influent concentration of 4.08 mg/L, was degraded to below detection limits (not specified) during a 20 hour aeration period in an aerobic activated sludge system(1). Camphor did not concentrate in the activated sludge solids. However, the loss of camphor in this process cannot be definitely attributed to biodegradation since there may have been loss due to volatilization(1). Monoterpene ketones were more resistant to biodegradation in aerated lagoon samples than monoterpene hydrocarbons or alcohols(2). The concentration of camphor actually increased in some of the samples which suggests that camphor was being produced in the lagoon. Camphor was on the list of very difficult to biodegrade compounds in a study of organic chemicals found in effluents(3). Camphor, present at 100 mg/L, reached 94% of its theoretical BOD in 4 weeks using an activated sludge inoculum at 30 mg/L in the Japanese MITI test which classified the compound as readily biodegradable(4).

12.3 Bioaccumulative potential

An estimated BCF of 17 was calculated in fish for camphor(SRC), using a measured log Kow of 2.38(1) and a regression-derived equation (2). According to a classification scheme(2), this BCF suggests the potential for bioconcentration in aquatic organisms is low(SRC).

12.4 Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc of camphor can be estimated to be 117(SRC). According to a classification scheme(2), this estimated Koc value suggests that camphor is expected to have high mobility in soil.

12.5 Other adverse effects

no data available

13. DISPOSAL CONSIDERATIONS

13.1 Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

14. TRANSPORT INFORMATION

14.1 UN Number

no data available

14.2 UN Proper Shipping Name

no data available

14.3 Transport hazard class(es)

no data available

14.4 Packing group, if applicable

no data available

14.5 Environmental hazards

no data available

14.6 Special precautions for user

no data available

14.7 Transport in bulk according to IMO instruments

no data available

15. REGULATORY INFORMATION

15.1 Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)	Listed.
EC Inventory	Listed.
United States Toxic Substances Control Act (TSCA) Inventory	Listed.
China Catalog of Hazardous chemicals 2015	Not Listed.
New Zealand Inventory of Chemicals (NZIoC)	Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)	Listed.
Vietnam National Chemical Inventory	Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)	Not Listed.
Korea Existing Chemicals List (KECL)	Listed.

16. OTHER INFORMATION

Information on revision

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Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

References

IPCS - The International Chemical Safety Cards (ICSC), website: <http://www.ilo.org/dyn/icsc/showcard.home>

HSDB - Hazardous Substances Data Bank, website: <https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm>

IARC - International Agency for Research on Cancer, website: <http://www.iarc.fr/>

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en

CAMEO Chemicals, website: <http://cameochemicals.noaa.gov/search/simple>

ChemIDplus, website: <http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp>

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: <http://www.phmsa.dot.gov/hazmat/library/erg>

Germany GESTIS-database on hazard substance, website: <http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp>

ECHA - European Chemicals Agency, website: <https://echa.europa.eu/>

Other Information

no data available

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