

According to the UN GHS revision 8

Creation Date: May 28, 2026

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## 1. IDENTIFICATION

### 1.1 GHS Product identifier

Product name: Chlorothalonil

Catalog Number: T19914

CAS Number: 1897-45-6

### 1.2 Other means of identification

Other names: -

### 1.3 Recommended use of the chemical and restrictions on use

Identified uses: no data available

### 1.4 Supplier's details

Company: Targetmol Chemicals Inc.

Address: 34 Washington Street, Wellesley Hills, Massachusetts 02481 USA

Tel/Fax: (781) 999-4286

### 1.5 Emergency phone number

Emergency phone number: 781-999-4286

Service hours: Monday to Friday, 9am-5pm (Standard timezone: UTC/GMT -5 hours).

## 2. HAZARD IDENTIFICATION

### 2.1 Classification of the substance or mixture

Acute toxicity, Inhalation (Category 2), H330

Serious eye damage (Category 1), H318

Skin sensitization (Category 1), H317

Carcinogenicity (Category 2), H351

Specific target organ toxicity - single exposure (Category 3), Respiratory system, H335

Short-term (acute) aquatic hazard (Category 1), H400

Long-term (chronic) aquatic hazard (Category 1), H410

For the full text of the H-Statements mentioned in this Section, see Section 16

### 2.2 GHS label elements, including precautionary statements

Pictogram(s):



Signal word: Danger

Hazard statement(s):

H317 May cause an allergic skin reaction.

H318 Causes serious eye damage.

H330 Fatal if inhaled.

H335 May cause respiratory irritation.

H351 Suspected of causing cancer.

H410 Very toxic to aquatic life with long lasting effects

Precautionary statement(s):

Prevention:

P201 Obtain special instructions before use.

P273 Avoid release to the environment.

P280 Wear protective gloves/ eye protection/ face protection

P302 + P352 IF ON SKIN: Wash with plenty of water.

P304 + P340 + P310 IF INHALED: Remove person to fresh air and keep comfortable for breathing. Immediately call a POISON CENTER/ doctor.

**Response:**

P305 + P351 + P338 +

P310

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER/ doctor.

**Storage:**

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

P405 Store locked up.

**Disposal:**

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

## 2.3 Other hazards which do not result in classification

This substance/mixture contains no components considered to be either persistent, bioaccumulative and toxic (PBT), or very persistent and very bioaccumulative (vPvB) at levels of 0.1% or higher.

## 3. COMPOSITION/INFORMATION ON INGREDIENTS

### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number
Chlorothalonil	-	1897-45-6	217-588-1

## 4. FIRST-AID MEASURES

### 4.1 Description of necessary first-aid measures

**General advice**

no data available

**If inhaled**

Fresh air, rest.

**Following skin contact**

Remove contaminated clothes. Rinse and then wash skin with water and soap. Refer for medical attention .

**Following eye contact**

Rinse with plenty of water (remove contact lenses if easily possible). Refer immediately for medical attention.

**Following ingestion**

Rinse mouth. Refer for medical attention .

### 4.2 Most important symptoms/effects, acute and delayed

1. Wash off dermal contamination with soap and water. Remove contamination of the eyes by flushing with copious amounts of water. If irritation persists, specialized medical care should be obtained. Substituted benzenes

### 4.3 Indication of immediate medical attention and special treatment needed, if necessary

SYMPTOMS: Symptoms of exposure to this compound include dermatitis and gastrointestinal, skin and upper respiratory tract irritation. ACUTE/CHRONIC HAZARDS: This compound is a positive animal carcinogen. When heated to decomposition it emits toxic fumes of chloride ion, NOx and cyanide ion. (NTP, 1992)

## 5. FIRE-FIGHTING MEASURES

### 5.1 Extinguishing media

Fire Extinguishing Media: CO<sub>2</sub>, foam, dry chemical or water.

### 5.2 Specific hazards arising from the chemical

Literature sources indicate that this compound is nonflammable. (NTP, 1992)

### 5.3 Special protective actions for fire-fighters

Use water spray, foam, powder, carbon dioxide.

## 6. ACCIDENTAL RELEASE MEASURES

### 6.1 Personal precautions, protective equipment and emergency procedures

Personal protection: chemical protection suit and protective gloves. Do NOT let this chemical enter the environment. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

### 6.2 Environmental precautions

Personal protection: chemical protection suit and protective gloves. Do NOT let this chemical enter the environment. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

### 6.3 Methods and materials for containment and cleaning up

Solid spillage should be picked up with an industrial vacuum cleaner and disposed of in accordance with local regulations.

## 7. HANDLING AND STORAGE

### 7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### 7.2 Conditions for safe storage, including any incompatibilities

Provision to contain effluent from fire extinguishing. Separated from food and feedstuffs. Store in an area without drain or sewer access. Keep in cool, dry, ventilated place.

Powder: -20°C for 3 years | In solvent: -80°C for 1 year

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

### 8.1 Control parameters

#### Occupational Exposure limit values

MAK sensitization of skin (SH)

#### Biological limit values

no data available

### 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

### 8.3 Individual protection measures, such as personal protective equipment (PPE)

#### Eye/face protection

Wear safety spectacles or eye protection in combination with breathing protection if powder.

#### Skin protection

Protective gloves. Protective clothing.

#### Respiratory protection

Use local exhaust or breathing protection.

#### Thermal hazards

no data available

## 9. PHYSICAL AND CHEMICAL PROPERTIES

Physical state                      Solid

Color                                      White

<b>Odour</b>	Odorless in pure form
<b>Melting point/freezing point</b>	251°C(lit.)
<b>Boiling point or initial boiling point and boiling range</b>	350°C(lit.)
<b>Flammability</b>	Liquid formulations containing organic solvents may be flammable. Gives off irritating or toxic fumes (or gases) in a fire.
<b>Lower and upper explosion limit/flammability limit</b>	no data available
<b>Flash point</b>	126°C(lit.)
<b>Auto-ignition temperature</b>	no data available
<b>Decomposition temperature</b>	no data available
<b>pH</b>	no data available
<b>Kinematic viscosity</b>	no data available
<b>Solubility</b>	DMSO: 17.86 mg/mL (67.17 mM),Sonication is recommended. (< 1 mg/ml refers to the product slightly soluble or insoluble)
<b>N-octanol-water partition coefficient</b>	log Kow = 3.05
<b>Vapour pressure</b>	4.36E-05mmHg at 25°C
<b>Density and/or relative density</b>	1.71g/cm3
<b>Relative vapour density</b>	no data available
<b>Particle characteristics</b>	no data available

## 10. STABILITY AND REACTIVITY

### 10.1 Reactivity

Decomposes on heating. This produces toxic and corrosive fumes including hydrogen chloride (see ICSC 0163) and nitrogen oxides.

### 10.2 Chemical stability

Thermally stable at ambient temperatures. Stable to u.v. light in aqueous media and in crystalline state. Stable in acidic and moderately alkaline aqueous solutions; slow hydrolysis at pH >9.

### 10.3 Possibility of hazardous reactions

Chlorothalonil is non-flammable and non-explosive. CHLOROTHALONIL is stable in neutral or acidic aqueous media. May react violently with strong oxidizing acids [Farm Chemicals Handbook]. Incompatible with other oxidizing agents such as peroxides and epoxides. Breaks down slowly in basic aqueous media (half-life 38.1 days at pH 9). [Farm Chemicals Handbook].

### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Cyanides

### 10.6 Hazardous decomposition products

May decompose at high temp to emit hydrogen chloride.

## 11. TOXICOLOGICAL INFORMATION

### Acute toxicity

Oral: LD50 Mouse oral 3700 mg/kg

Inhalation: LC50 Rat inhalation 310 mg/cu m/1hr

Dermal: LD50 Rabbit (albino) percutaneous >5000 mg/kg

### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

### Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

### Carcinogenicity

Cancer Classification: Group B2 Probable Human Carcinogen

### Reproductive toxicity

no data available

### STOT-single exposure

The substance is severely irritating to the eyes. The substance is irritating to the respiratory tract. The substance is mildly irritating to the skin.

### STOT-repeated exposure

Repeated or prolonged contact with skin may cause dermatitis. Repeated or prolonged contact may cause skin sensitization.

### Aspiration hazard

A harmful concentration of airborne particles can be reached quickly when dispersed.

## 12. ECOLOGICAL INFORMATION

### 12.1 Toxicity

Toxicity to fish: LC50; Species: /Oncorhynchus mykiss/ (Rainbow trout); Concentration: 76 ug/L for 96 hr /Technical chlorothalonil; conditions of bioassay not specified

Toxicity to daphnia and other aquatic invertebrates: EC50; Species: Daphnia magna (Water flea) age <24 hr; Conditions: static; Concentration: 70 ppb for 48 hr (95% confidence limit: 34.2-143 ppb); Effect: intoxication, immobilization) /96% AI formulated product

Toxicity to algae: EC50; Species: Pseudokirchneriella subcapitata (Green algae); Conditions: static; Concentration: 190 ppb/120 hr (95% confidence limit: 180-210 ppb); Effect: decreased population abundance /97.9% AI formulated product

Toxicity to microorganisms: no data available

### 12.2 Persistence and degradability

AEROBIC: In an aerobic grab sample study, 60.5% chlorothalonil (initial concentration of 38 ppm) remained in an alluvial silty loam (25 deg C and pH 6.4) after 7 days(1). At an initial concentration of 40 ppm, 67.5% (20 deg C) and 45.0% (25 deg C) chlorothalonil remained after 7 days in an alluvial silty loam (pH 6.4) held at a 60% moisture capacity(1). In the same soil held at moisture capacities of 100, 60, 40, and 20%, the amount of chlorothalonil (initial concentration of 40 ppm) remaining was determined to be 85, 22, 45, and 65%, respectively, after 7 days. Chlorothalonil biodegraded mainly through dechlorination and partly by substitution reaction yielding the biodegradation products: isophthalonitrile, mono-, di- and tri-chlorinated isophthalonitriles, 2,5,6-trichloro-4-hydroxyisophthalonitrile and 2,5,6-trichloro-4-methoxyisophthalonitrile(1). Aerobic biodegradation half-lives of chlorothalonil in four different soils were reported as 10, 10, 15 and 40 days(2). After 60 days the metabolite 2,5,6-trichloro-4-hydroxyisophthalonitrile was present at up to 32% of the initially applied amount; the metabolite 3-cyano-2,4,5,6-tetrachlorobenzamide was present at up to about 7% at both days 7 and 16 of the study. Aerobic aquatic half-lives of chlorothalonil typically range from 2 to 200 hours(2). Other reported metabolites are 1,3-dicarbamoyl-2,4,5,6-tetrachlorobenzene, 1-carbamoyl-3-cyano-4-hydroxy-2,5,6-trichlorobenzene, 2,4,6-trichloro-4-(methylthio) isophthalonitrile and pentachloronitrobenzene(3).

### 12.3 Bioaccumulative potential

Bioconcentration factors of 75 (edible) and 264 (whole body) were measured for chlorothalonil in bluegill sunfish(1). BCF values of 9.4 (edible) and 16 (whole body) were reported for catfish(1). Carp exposed to 3 ug/L chlorothalonil over a 6 week incubation period had reported maximum BCF values of 125 (2). A BCF of 264 was determined in bluegill (*Lepomis macrochirus*) exposed for 28 days(3). According to a classification scheme(4), these BCF values suggest bioconcentration in aquatic organisms can be low to high(SRC). Chlorothalonil BAFs wet weight and lipid based in zooplankton from three lakes in southern Ontario, Canada were 14-1500 and 140-190,000, respectively; samples were collected in 2003 and 2004(5).

### 12.4 Mobility in soil

A Koc value of 1800 has been determined based on adsorption isotherms of chlorothalonil on 3 black soils and 1 clay mineral, Na-bentonite soil(1). Other reported Koc values are 1300 to 14,000(2). Laboratory batch equilibrium studies with four soils showed chlorothalonil to possess limited mobility in silty clay loam, silt, sandy loam, and sand with calculated Koc values were 1400, 7000, 1100,

## A DRUG SCREENING EXPERT

and 900, respectively(3). According to a classification scheme(4), these Koc values suggest the mobility of chlorothalonil in soils will be low to immobile(SRC). The Koc of sediment from recycling ponds was 2270 to 2450(5).

### 12.5 Other adverse effects

no data available

## 13. DISPOSAL CONSIDERATIONS

### 13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

#### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

## 14. TRANSPORT INFORMATION

### 14.1 UN Number

no data available

### 14.2 UN Proper Shipping Name

no data available

### 14.3 Transport hazard class(es)

no data available

### 14.4 Packing group, if applicable

no data available

### 14.5 Environmental hazards

no data available

### 14.6 Special precautions for user

no data available

### 14.7 Transport in bulk according to IMO instruments

no data available

## 15. REGULATORY INFORMATION

### 15.1 Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)	Listed.
EC Inventory	Listed.
United States Toxic Substances Control Act (TSCA) Inventory	Listed.
China Catalog of Hazardous chemicals 2015	Not Listed.
New Zealand Inventory of Chemicals (NZIoC)	Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)	Listed.
Vietnam National Chemical Inventory	Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)	Listed.
Korea Existing Chemicals List (KECL)	Listed.

## 16. OTHER INFORMATION

## Information on revision

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## Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

## References

IPCS - The International Chemical Safety Cards (ICSC), website: <http://www.ilo.org/dyn/icsc/showcard.home>

HSDB - Hazardous Substances Data Bank, website: <https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm>

IARC - International Agency for Research on Cancer, website: <http://www.iarc.fr/>

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: [http://www.echemportal.org/echemportal/index?pageID=0&request\\_locale=en](http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en)

CAMEO Chemicals, website: <http://cameochemicals.noaa.gov/search/simple>

ChemIDplus, website: <http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp>

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: <http://www.phmsa.dot.gov/hazmat/library/erg>

Germany GESTIS-database on hazard substance, website: <http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp>

ECHA - European Chemicals Agency, website: <https://echa.europa.eu/>

## Other Information

Carrier solvents used in commercial formulations may change physical and toxicological properties.

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*Disclaimer: The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. We as supplier shall not be held liable for any damage resulting from handling or from contact with the above product. All products are for Research Use Only · Not For Human or Veterinary or Therapeutic Use*