

2,2-Dihydroxyacetic acid

Chemical Properties

CAS No. : 563-96-2

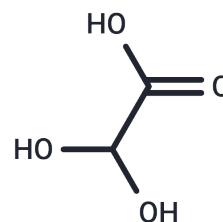
Formula: C₂H₄O₄

Molecular Weight: 92.05

Keep away from moisture

Storage: Powder: -20°C for 3 years | In solvent: -80°C for 1 year

Actual storage temperature shall be subject to the COA.



Biological Description

| | |
|---------------|--|
| Description | 2,2-Dihydroxyacetic acid (Formylformic acid) is an intermediate in the glyoxalate cycle, which enables certain organisms to convert fatty acids to carbohydrates. 2,2-Dihydroxyacetic acid has been found to be associated with primary hyperoxaluria, an inborn error of metabolism. As an aldehyde, glyoxalate is also highly active and can modify proteins to form advanced glycosylation products (AGEs). |
| Targets(IC50) | Endogenous Metabolite |

Solubility Information

| | |
|---------------------|--|
| Solubility | DMSO: 247.5 mg/mL (2688.76 mM), Sonication is recommended. (< 1 mg/ml refers to the product slightly soluble or insoluble) |
| In vivo Formulation | 10% DMSO+40% PEG300+5% Tween 80+45% Saline: 2 mg/mL (21.73 mM), Sonication is recommended. <i>Please add the solvents sequentially, clarifying the solution as much as possible before adding the next one. Dissolve by heating and/or sonication if necessary. Working solution is recommended to be prepared and used immediately. The formulation provided above is for reference purposes only. In vivo formulations may vary and should be modified based on specific experimental conditions.</i> |

Preparing Stock Solutions

| | 1mg | 5mg | 10mg |
|-------|------------|------------|-------------|
| 1 mM | 10.8637 mL | 54.3183 mL | 108.6366 mL |
| 5 mM | 2.1727 mL | 10.8637 mL | 21.7273 mL |
| 10 mM | 1.0864 mL | 5.4318 mL | 10.8637 mL |
| 50 mM | 0.2173 mL | 1.0864 mL | 2.1727 mL |

Please select the appropriate solvent to prepare the stock solution, according to the solubility of the product in different solvents. Please use it as soon as possible.

Note: The dilution table applies only to solid products. For liquid products, please calculate the stock solution based on the stated concentration and/or density.

Reference

Lee S H , Kim S O , Chung B C . Gas chromatographic-mass spectrometric determination of urinary oxoacids using O-(2,3,4,5,6-pentafluorobenzyl)oxime-trimethylsilyl ester derivatization and cation-exchange chromatography[J]. Journal of Chromatography B, 1998, 719(1-2):1-7.

Wang Z, Cui L, Lin Y, et al. Cancer cell-intrinsic biosynthesis of itaconate promotes tumor immunogenicity. The EMBO Journal. 2024: 1-18.

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