

2-Hexylthiophene

Chemical Properties

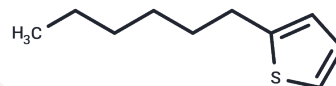
CAS No. : 18794-77-9

Formula: C₁₀H₁₆S

Molecular Weight: 168.3

Storage: Powder: -20°C for 3 years | In solvent: -80°C for 1 year

Actual storage temperature shall be subject to the COA.



Biological Description

Description	2-Hexylthiophene, a weakly basic heteroaromatic compound, can enhance the molar absorption coefficient of ruthenium sensitizers by undergoing modification.
Targets(IC50)	Others
In vitro	By incorporating 2-Hexylthiophene at the 5-position of the terpyridine derivative ligand, an enhancement in the molar absorption coefficient of the ruthenium sensitizer is achieved.[1]

Solubility Information

Solubility	DMSO: 90 mg/mL (534.76 mM),Sonication is recommended. (< 1 mg/ml refers to the product slightly soluble or insoluble)
In vivo Formulation	10% DMSO+40% PEG300+5% Tween 80+45% Saline: 3.3 mg/mL (19.61 mM),Sonication is recommended. <i>Please add the solvents sequentially, clarifying the solution as much as possible before adding the next one. Dissolve by heating and/or sonication if necessary. Working solution is recommended to be prepared and used immediately. The formulation provided above is for reference purposes only. In vivo formulations may vary and should be modified based on specific experimental conditions.</i>

Preparing Stock Solutions

	1mg	5mg	10mg
1 mM	5.9418 mL	29.7089 mL	59.4177 mL
5 mM	1.1884 mL	5.9418 mL	11.8835 mL
10 mM	0.5942 mL	2.9709 mL	5.9418 mL
50 mM	0.1188 mL	0.5942 mL	1.1884 mL

Please select the appropriate solvent to prepare the stock solution, according to the solubility of the product in different solvents. Please use it as soon as possible.

Note: The dilution table applies only to solid products. For liquid products, please calculate the stock solution based on the stated concentration and/or density.

Reference

Ozawa H, et al. Ruthenium sensitizers with a hexylthiophene-modified terpyridine ligand for dye-sensitized solar cells: synthesis, photo- and electrochemical properties, and adsorption behavior to the TiO₂ surface. *ACS Appl Mater Interfaces*. 2015 Feb 11;7(5):3152-61.

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